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Types Of Reactions Worksheet Answer

Types of Reactions Worksheet -
Solutions 1) $3 \text{ NaBr} + 1 \text{ H}_3 \text{ PO}_4 + 1 \text{ Na}_3$

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PO₄ + 3 HBr Reaction type: _____ double displacement
2) 3 Ca(OH)₂ + 1 Al₂(SO₄)₃ Reaction type: _____ double displacement
3) 3 Mg + 1 Fe₂O₃ Reaction type: _____ double displacement
4) 3 CaSO₄ + 2 Al(OH)₃ Reaction type: _____ double displacement
5) 2 Fe + 3 MgO Reaction type: _____ double displacement

Types of Reactions Worksheet

Types of Reactions Worksheet -
Solutions Balance the following

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equations and indicate the type of reaction taking place: 1) $3 \text{NaBr} + 1 \text{H}_3\text{PO}_4 \rightarrow 1 \text{Na}_3\text{PO}_4 + 3 \text{HBr}$ Type of reaction: double displacement 2) $3 \text{Ca(OH)}_2 + 1 \text{Al}_2(\text{SO}_4)_3 \rightarrow 3 \text{CaSO}_4 + 2 \text{Al(OH)}_3$ Type of reaction: double displacement 3) $3 \text{Mg} + 1 \text{Fe}_2\text{O}_3 \rightarrow 2 \text{Fe} + 3 \text{MgO}$ Type of reaction: single displacement

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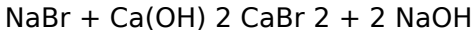
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main things we will present to you based
on the ...

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Types of Chemical Reactions Answers

Balance each of the following reactions
and identify each type of reaction: 1. 2



double displacement 2. 2 $\text{NH}_3 + \text{H}_2\text{SO}_4$

4 $(\text{NH}_4)_2\text{SO}_4$ synthesis 3. 4 $\text{C}_5\text{H}_9\text{O}$

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+ 29 O₂ 20 CO₂ + 18 H₂O

combustion 4. 3 Pb + 2 H₃PO₄ 3 H₂ +

Pb₃(PO₄)₂ single displacement 5. Li₃

N + 3 NH₄NO₃ 3 LiNO₃ + (NH₄)₃

Types of Chemical Reaction Worksheet

Five Types of Chemical Reaction
Worksheet Answers. Balance the

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following reactions and indicate which of the six types of chemical reaction are being represented: a) $2. \text{NaBr} + 1. \text{Ca(OH)}_2 \rightarrow 1. \text{CaBr}_2 + 2. \text{NaOH}$ Type of reaction: double displacement. b) $2. \text{NH}_3 + 1. \text{H}_2\text{SO}_4 \rightarrow 1. (\text{NH}_4)_2\text{SO}_4$ Type of reaction: synthesis. c) $4. \text{C}_5\text{H}_9\text{O} + 27. \text{O}_2 \rightarrow 20. \text{CO}_2 + 18. \text{H}_2\text{O}$ Type of reaction:

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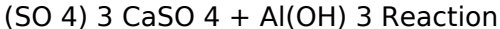
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Types of Reactions Worksheet Indicate the type of reaction taking place: 1)

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Reaction type: _____ 2) $\text{Ca(OH)}_2 + \text{Al}_2$



Reaction type: _____ 3) $\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow \text{Fe} + \text{MgO}$

Reaction type: _____ 4) $\text{C}_2\text{H}_4 + \text{O}_2$

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Types of Reactions Worksheet -

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Solutions. Balance the following equations and indicate the type of reaction taking place: 1) $3. \text{NaBr} + 1. \text{H}_3\text{PO}_4 \rightarrow 1. \text{Na}_3\text{PO}_4 + 3. \text{HBr}$. Type of reaction: double displacement. 2) $3. \text{Ca}(\text{OH})_2 + 1. \text{Al}_2(\text{SO}_4)_3 \rightarrow 3. \text{CaSO}_4 + 2. \text{Al}(\text{OH})_3$. Type of reaction: double displacement. 3) $3. \text{Mg} + 1. \text{Fe}_2\text{O}_3 \rightarrow 2. \text{Fe} + 3. \text{MgO}$. Type of ...

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Types of Chemical Reaction Worksheet.
Balance the reactions 1 to 6. and
indicate which type of chemical reaction
(synthesis, decomposition, single-
displacement, double-displacement or
combustion) is being represented: 2_

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NaBr + 1 Ca(OH)₂ (1 CaBr₂ + 2
NaOH Reaction Type : Double
replacement. 2 NH₃ + 1 H₂SO₄ (1
(NH₄)₂SO₄ Reaction Type : Synthesis.

Types of Chemical Reaction Worksheet

Types of Reactions Worksheet -
Solutions. Balance the following

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equations and indicate the type of reaction taking place: 1) $3. \text{NaBr} + 1. \text{H}_3\text{PO}_4 \rightarrow 1. \text{Na}_3\text{PO}_4 + 3. \text{HBr}$ Type of reaction: ... - Answers. Balance these equations! Note to students: Whenever balancing an equation, it is acceptable to leave spaces blank instead of writing "1 ..."

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Balancing Equations Practice Worksheet

Six Types of Chemical Reaction
Worksheet Answers Balance the
following reactions and indicate which of
the six types of chemical reaction are
being represented: a) $2 \text{NaBr} + 1$
 $\text{Ca(OH)}_2 \rightarrow 1 \text{CaBr}_2 + 2 \text{NaOH}$

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Six Types of Chemical Reaction
Worksheet Answers Balance the
following reactions and indicate which of
the six types of chemical reaction are
being represented: a) $2 \text{NaBr} + 1 \text{Ca} \dots$

Balance the reactions a to e and

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indicate which types of ...

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